

Electron transfer reactions in photosystem I following vitamin K₁ depletion by ultraviolet irradiation

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Photosystem I preparations were irradiated with UV to destroy vitamin K₁ in situ. The depletion of vitamin K₁ resulted in inactivation of NADP⁺ photoreduction and introduction of a ~220 ms component in the flash generated P700⁺ re-reduction at room temperature. The photoreduction of the terminal FeS centers FA and FB in control and vitamin K₁-depleted preparations at 7 K were comparable. The data confirm that vitamin K₁ is functionally implicated in primary electron transfer reactions in PS I at physiological temperature, and that the anomalous results at cryogenic temperature may be explicable in terms of a by-pass of the vitamin K₁ acceptor site or heterogeneity introduced into the photosystem by quinone removal.

Photosystem I; Primary reaction; Vitamin K₁; Ultraviolet irradiation; EPR; Optical transient

1. INTRODUCTION

The primary charge separation in PS I results in the formation of the radical pair P700⁺A₀⁻. Subsequent rapid electron transfer reactions occur through secondary redox centers referred to as A₁, Fx and FA/FB. The terminal acceptors FA/FB have been identified as two 4Fe:4S clusters associated with an 8.9 kDa polypeptide localized on the stromal domain of the PS I core to interface with the enzymatic apparatus for NADP⁺ photoreduction. Fx is more electronegative than FA/FB and is most likely another 4Fe:4S cluster, but is associated with the PS I core which comprises the 82 and 83 kDa polypeptides of the PS I reaction center (see [1] for review).

The identity of the acceptor A₁, the postulated donor to Fx, is presently controversial. A₁ was pro-

visionally identified as vitamin K₁ based upon a consideration of its EPR signal [2,3] and, more recently, a comparison of the electron spin polarized flash-induced EPR K-band spectrum of P700⁺A₁⁻ with that of P870⁺Q⁻ in Fe-depleted bacterial reaction centers [4]. Direct support for the participation of vitamin K₁ was provided by Brettel et al. [5] following a detailed examination of optical flash transients in UV at cryogenic temperature. Subsequently positive identification of the junctional involvement of vitamin K₁ was obtained by solvent extraction of vitamin K₁ and the specific reconstitution of room temperature biochemical activities of PS I using exogenous phyloquinone [6]. These reconstitution studies have recently been confirmed using an ether extracted preparation of PS I from spinach [7].

Examination of the solvent-extracted PS I preparations that were totally depleted of vitamin K₁ at cryogenic temperature led to anomalous results in that the centers FA/FB were found to be phototrapped with high efficiency [8] and the EPR signal corresponding to A₁ was still observed (Setif, P., personal communication). Additional controversy regarding the participation of vitamin K₁

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Abbreviations: DPIP, dichlorophenolindophenol; P700, primary electron donor of photosystem I; A₀, primary electron acceptor of photosystem I; Chl, chlorophyll

in PS I electron transfer came from studies of preparations of PS I that were depleted of vitamin K₁ by UV irradiation to destroy the phyloquinone in situ. Palace et al. [9] reported that the kinetics of the room temperature charge recombination between P700⁺ and the reduced terminal FeS centers was not affected following vitamin K₁ destruction suggesting the non-participation of vitamin K₁ in the electron transfer sequence. Zeigler et al. [10] also showed that depletion of vitamin K₁ by irradiation treatment of PS I did not alter the EPR spectrum of A₁ and concluded that the signal ascribed to the acceptor A₁ does not originate from vitamin K₁.

In efforts to resolve these discrepancies we have reinvestigated the effect of UV irradiation on the electron transfer activities of PS I at physiological temperature and included a study of the behavior of FeS redox centers at 7 K on the same preparations. The data are consistent with the requirement for vitamin K₁ for the photoreduction of NADP⁺ and that it is an electron transfer component earlier than Fx in confirmation of [6]. We also confirm and extend the optical flash transient studies of P700⁺ recombination reactions at room temperature [9] and EPR studies of reduced FeS centers at cryogenic temperature [8,10].

2. MATERIALS AND METHODS

2.1. Preparations

Spinach D-144 preparations were isolated according to Anderson and Boardman [11] and stored at -80°C. The PS I preparations had an antenna size of ~290 Chl *a,b*/P700 and a Chl *a/b* ratio of 4.38 ± 0.29.

2.2. UV irradiation

The UV irradiation of PS I was conducted at 2°C in a Pyrex thermostatted chamber at a Chl concentration of 10 µg/ml. The light source, Blak Ray model B100A (Ultraviolet Products, San Gabriel, CA) provided wavelengths of 300–400 nm at an intensity of 50 W · m⁻². These conditions were similar to those used by Palace et al. [9].

2.3. Biochemical assays

Vitamin K₁ was extracted and analyzed using reversed-phase HPLC [6] except that the eluant was methylene chloride/methanol (10:90). NADP⁺ photoreduction was determined spectrophotometrically at 340 nm using supplementary ferredoxin and ferredoxin-NADP⁺-reductase (Sigma Chemical Co., St Louis, MO) and 20 mM ascorbate and 100 µM DPFP as donor. The reactions were conducted at light saturation but they were most probably limited by the absence of plastocyanin as physiological donor.

2.4. Optical flash transients

Flash-induced absorbance transients were detected at both 700 and 820 nm using a PIN 10D silicon photodiode (United Detector Technology, Hawthorne, CA). The detector output was amplified (model 113 preamp, EG&G/PARC, Princeton, NJ) and the transients were averaged using a 125 MHz digital oscilloscope (model 9400, LeCroy, Chestnut Ridge, NY). Saturating activation flashes were provided by a Q-switched, frequency doubled Nd:YAG laser (model DCR 11, Spectra Physics, Mountain View, CA). The flashes were 532 nm and 8–9 ns (full width at half height).

2.5. EPR spectroscopy

EPR experiments were carried out using an X-band spectrometer (Varian Associates, Palo Alto, CA) employing an E102-E bridge, with field modulation frequency, 100 kHz; field modulation amplitude, 16 G; lock-in sensitivity, 100 mV; sweep time, 10 min; time constant, 1 s; microwave frequency, 9.055 GHz, and microwave power, 50 mW. The temperature was maintained between 7–10 K using a helium flow cryostat (Air Products, Allentown, PA).

3. RESULTS

The spinach D-144 PS I preparations were analyzed and found to contain 2.12 ± 0.38 mol vitamin K₁/mol P700. Upon irradiation using wavelengths in the range 300–400 nm the vitamin K₁ was totally depleted after 2–2.5 h. Fig.1 shows the flash-induced absorbance signal at 700 nm due to P700⁺. The decay halftime for the optical transient in the control preparation was ~ 30 ms and can be assumed to be due to the recombination between P700⁺ and the reduced terminal acceptors FA/FB [1]. After vitamin K₁ depletion (lower trace), the optical transient still showed a slow recombination in confirmation of [9] but the decay halftime was longer. Fig.2 (upper panel) shows that, during the irradiation treatment, the decay halftime of the transient increased to ~ 60 ms, and upon more detailed analysis, we found that this was due to the introduction of a slower decay component with a halftime of approx. 220 ms. This contrasts with Palace et al. [9] who reported that the vitamin K₁ depletion did not modify the P700 recombination kinetics.

Fig.2 (lower panel) shows that the inactivation of NADP⁺ photoreduction also closely followed the depletion of vitamin K₁ during irradiation treatment. These results confirm those observed following solvent extraction of *Synechocystis* PS I [6], and correlate the loss of physiological electron transfer function with vitamin K₁ removal.

The behavior of the terminal FeS redox centers

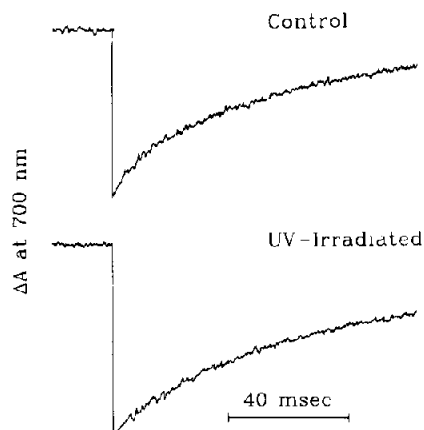


Fig.1. Flash absorption transients due to $P700^+$ in PS I (control) and following vitamin K_1 depletion (UV-irradiated). The transient amplitudes (ΔA) for the control and UV-irradiated samples were 1.66×10^{-3} and 1.72×10^{-3} , respectively.

FA/FB was investigated via EPR at cryogenic temperatures for control and irradiated samples. The samples were dark-adapted at room temperature in the presence of ascorbate and DPIP and cooled in the dark to ~ 7 K. After recording a dark baseline, which was used for the correction of subsequent spectra, the samples were illuminated for 15 s and then, following a short dark period, the spectra were recorded in darkness to show irreversibly reduced redox centers (phototrapped). The upper panel of Fig.3 shows

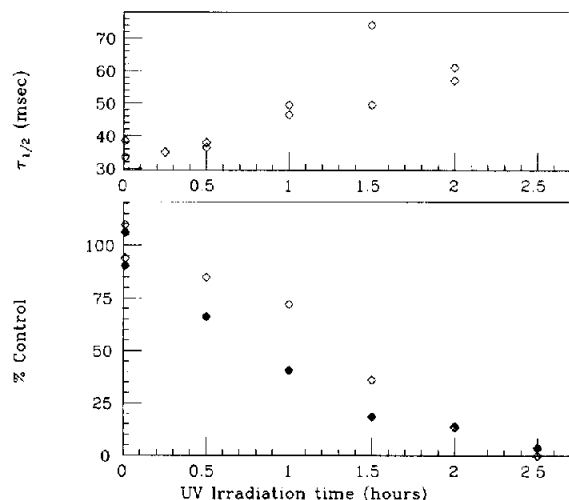


Fig.2. (Upper panel) The effect of UV irradiation on the half-time of the $P700^+$ recovery following flash activation at room temperature. (Lower panel) The effect of UV irradiation on vitamin K_1 concentration (\blacklozenge) and $NADP^+$ photoreduction (\circ).

that the FeS center FA was phototrapped in both the control and the sample depleted of vitamin K_1 (UV irradiated). The samples were then illuminated for several minutes, and then EPR spectra were determined in continuous light. In continuous light (lower panel of Fig.3) center FB was resolved as well as center FA for both samples. In both the phototrapped and continuous light treatments, the EPR intensities arising from the FA/FB centers in control and UV-irradiated samples were com-

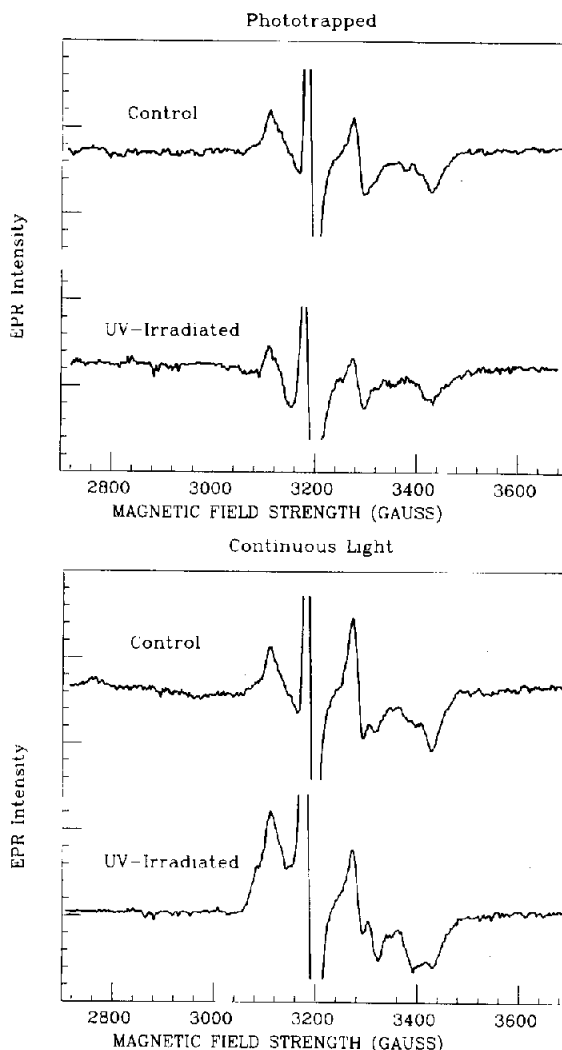


Fig.3. EPR spectra obtained between 7 and 10 K of control and UV-irradiated spinach PS I preparations. (Upper panel) Samples dark adapted at room temperature, cooled in darkness and then illuminated for 15 s. The spectra show redox centers that were phototrapped. (Lower panel) Spectra acquired during continuous illumination of the samples.

parable indicating that the terminal FeS clusters were photoreduced at cryogenic temperature in PS I preparations depleted of vitamin K₁.

4. DISCUSSION

The data presented above on the electron transfer activities of PS I preparations depleted of vitamin K₁ by UV destruction in situ are similar to those reported previously for preparations treated with organic solvents to extract the vitamin K₁ [6,8]. In both protocols for vitamin K₁ removal, we observed the loss of NADP⁺ photoreduction at physiological temperature [6], but photoreduction of the centers FA and FB at cryogenic temperature was not impaired [8].

An important difference shown in this study was the room temperature flash absorbance transient due to the charge recombination between reduced acceptor(s) and P700⁺. In solvent extracted PS I preparations from *Synechocystis* we showed that the 30 ms charge recombination reaction between P700⁺ and reduced FeS was replaced by a much faster reaction [6]. This was interpreted as a recombination between partners of the primary biradical P700⁺A₀⁻, and decay of triplet P700 formed from this species. This process has been shown to be reversible upon addition of vitamin K₁ in a reconstitution process that has recently been confirmed and extended using an ether extracted spinach PS I preparation [7]. This behavior contrasts with the UV-irradiated PS I preparations presented here where although we confirm [9] that vitamin K₁ destruction in situ does not lead to the formation of an ultra-fast recombination reaction at the expense of the slow recombination of P700⁺P430⁻, the decay halftime increased during vitamin K₁ depletion in preparations that were also undergoing inactivation of NADP⁺ photoreduction. The increase in decay halftime was found to be due to the introduction of a much slower recombination pathway at the expense of the 30 ms component showing that, in contrast to [9], the P700⁺ reduction kinetics are extensively modified following vitamin K₁ removal.

We conclude that the similarity between the physiological electron transfer activities of PS I preparations depleted of vitamin K₁ via in situ destruction with those of solvent extracted preparations [6,7] confirms that vitamin K₁ is a

participant in the electron transfer pathway for NADP⁺ photoreduction. It was previously suggested that the anomalous data obtained at cryogenic temperature was explicable if either a bypass of the vitamin K₁ acceptor site occurred following vitamin K₁ removal, or some other heterogeneity in the pathway had been introduced [6,8]. The flash transient data presented above show that some heterogeneity in electron transfer is also observed at room temperature following vitamin K₁ destruction via UV irradiation.

Finally, we suggest that although the extraction and reconstitution data appear to implicate a functional role for vitamin K₁ on the acceptor side of PS I, vitamin K₁ is most likely not the chemical identity of the EPR signal generally referred to as A₁. Both solvent-extracted and UV-irradiated PS I preparations with no detectable vitamin K₁ continue to show the EPR signal characteristic of A₁ ([10] and Setif, personal communication) and the photoaccumulated EPR spectra in the PS I of cyanobacteria containing protonated and deuterated vitamin K₁ (position 2-methyl group) are identical indicating that the radical does not originate from phylloquinone [12].

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REFERENCES

- [1] Goldbeck, J.H. (1989) *Biochim. Biophys. Acta*, in press.
- [2] Bonnerjea, J. and Evans, M.C.W. (1982) *FEBS Lett.* 148, 313-316.
- [3] Gast, P., Swarthoff, T., Ebskamp, F.C.R. and Hoff, A.J. (1983) *Biochim. Biophys. Acta* 722, 168-175.
- [4] Peterson, J., Stehlick, D., Gast, P. and Thurnauer, M. (1987) *Photosynth. Res.* 14, 15-29.
- [5] Brettel, K., Setif, P. and Mathis, P. (1987) *FEBS Lett.* 203, 220-224.
- [6] Biggins, J. and Mathis, P. (1988) *Biochemistry* 27, 1494-1500.
- [7] Itoh, S. and Iwaki, M. (1989) *FEBS Lett.* 243, 47-52.
- [8] Setif, P., Ikegami, I. and Biggins, J. (1987) *Biochim. Biophys. Acta* 894, 146-156.
- [9] Palace, G.P., Franke, J.E. and Warden, J.T. (1987) *FEBS Lett.* 215, 58-62.
- [10] Zeigler, K., Lockau, W. and Nitschke, W. (1987) *FEBS Lett.* 217, 16-20.
- [11] Anderson, J.M. and Boardman, N.K. (1966) *Biochim. Biophys. Acta* 112, 403-421.
- [12] Barry, B., Bender, C.J., McIntosh, L., Ferguson-Miller, S. and Babcock, G.T. (1989) *Isr. J. Chem.*, in press.